



(IIb) (m.p. 127–130° from MeOH) which by reduction with LiAlH_4 and subsequent treatment with aqueous HCl gave III isolated as chlorohydrate (m.p. > 300°, $[\alpha]_D^{25} + 19.6^\circ$ in MeOH)⁴.

These 3 steps could be carried out without the purification of the intermediary products in an overall yield of 50%.

In an alternative route we converted the ditosylate Ia into the corresponding 20, 20-ethylenedioxy derivative Ib⁵ and then, following the general procedure indicated above, into the azido compound IIb in 80% yield.

The elemental analyses data as well as the IR-spectra are in perfect agreement with the formulae assigned to the described compounds.

Résumé. L'holaphyllamine peut être aisément préparée à partir du 3 α , 6 α -ditosyloxy-5 β -pregnan-20-one par réaction avec NaN_3 .

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⁴ The authors are indebted to Dr. R. GOUTAREL (Institut de Chimie des Substances Naturelles, Gif-sur-Yvette) under whose leadership the comparison of this substance with an authentic sample of holaphyllamine chlorohydrate was performed.

⁵ K. R. BHARUCHA, G. C. BUCKLEY, C. K. CROSS, L. J. RUBIN and P. ZIEGLER, Can. J. Chem. 34, 982 (1956).

Chemical Investigation of *Pluchea lanceolata* II. Identity of Pluchine with Betaine Hydrochloride

In a recent communication¹, the isolation of a quaternary base chloride named pluchine, m.p. 243–244° (decomposition and volatilization with evolution of gas), $[\alpha]_D^{20} - 29.51$ (H_2O) was reported from the whole plant of *Pluchea lanceolata*, Linn. (N.O. Compositae). Pluchine has since been identified as betaine hydrochloride by comparison of its paper chromatographic behaviour, m.p. and mixed m.p. with an authentic specimen.

Analysis found for pluchine: C, 38.74, 38.59, 39.12, 39.09; H, 7.79, 7.67, 7.78, 7.81; N, 8.64, 8.30, 7.49, 7.68; Cl, 23.55, 23.80, 22.85, 22.80. Analysis required for betaine hydrochloride, Cl^- , $\text{N}^+(\text{CH}_3)_3 \cdot \text{CH}_2 \cdot \text{COOH}$: C, 39.08; H, 7.81; N, 9.12; Cl, 23.12.

Both pluchine picrate and the picrate prepared from betaine hydrochloride melted at 181–182°, and there was no depression in their mixed m.p. Analysis found for pluchine picrate: C, 38.46, 38.58; H, 3.87, 4.08; N, 16.09, 15.86. Analysis required for betaine picrate, $(\text{NO}_2)_3 \text{C}_6\text{H}_5\text{O}^- \text{N}^+(\text{CH}_3)_3 \cdot \text{CH}_2 \cdot \text{COOH}$: C, 38.15; H, 4.04; N, 16.18.

Thus, the analyses of pluchine and its picrate agree well with those for betaine hydrochloride and its picrate.

PRASAD et al.² found pluchine to have anti-inflammatory and anti-arthritis activity in experimental animals (albino rats) with inflammation produced by both immunological and non-immunological methods. Pluchine was found not only to suppress the acute inflammation induced by carrageenin, histamine and formaldehyde in comparison with betamethasone, a known anti-inflammatory agent, but also effectively to suppress both primary and secondary phases of adjuvant arthritis (induced by

a suspension of dead tubercle bacilli, human DT strain, in liquid paraffin) like metamethasone. Pluchine significantly suppressed acute sensitivity reaction produced by purified tuberculin, and was less toxic than betamethasone in albino rats. These interesting properties of pluchine or betaine hydrochloride can account for the use of the drug, *Pluchea lanceolata*, for the treatment of rheumatism in clinical cases in the ayurvedic system of medicine.

Zusammenfassung. Die Struktur von Pluchin, einer quaternären Base aus *Pluchea lanceolata*, wurde als die des Betainhydrochlorids aufgeklärt. Damit sind auch die mit Pluchin beobachteten biologischen Wirkungen verständlich.

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¹ B. DASGUPTA, Experientia 23, 989 (1967).

² D. N. PRASAD, G. V. SATYAWATI, B. N. DASGUPTA and P. K. DAS, Abstracts of papers, First Congress of the South East Asia and Pacific Area League against Rheumatism, Bombay, India, 17–21 February 1968, p. 88.